

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR
(AUTONOMOUS)

B.Tech. I Year II Semester Regular & Supplementary Examinations June-2025
CHEMISTRY

(Common to CSE, CSIT, EEE & ECE)

Time: 3 Hours

Max. Marks: 70

PART-A

(Answer all the Questions 10 x 2 = 20 Marks)

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|---|---|--|-----|----|----|
| 1 | a | Define HOMO and LUMO. | CO1 | L1 | 2M |
| | b | Give the Significance of Ψ and Ψ^2 . | CO1 | L1 | 2M |
| | c | Mention the Properties of Nano materials. | CO2 | L1 | 2M |
| | d | What are Intrinsic and Extrinsic Semiconductors? | CO2 | L1 | 2M |
| | e | Give any two applications of Hydrogen Oxygen fuel cells. | CO3 | L1 | 2M |
| | f | Give examples for amperometric sensors. | CO4 | L1 | 2M |
| | g | What is monomer? | CO5 | L1 | 2M |
| | h | Give examples of Biodegradable polymers used in real life. | CO5 | L1 | 2M |
| | i | What is a stationary phase? | CO6 | L1 | 2M |
| | j | Define Beer- Lambert's law with equation. | CO6 | L1 | 2M |

PART-B

(Answer all Five Units 5 x 10 = 50 Marks)

UNIT-I

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|---|--|---|-----|----|-----|
| 2 | | Derive equation for a particle in one dimensional box in detail and explain their significance. | CO1 | L2 | 10M |
|---|--|---|-----|----|-----|

OR

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|---|---|---|-----|----|----|
| 3 | a | Explain Planck's Quantum Theory. | CO2 | L2 | 5M |
| | b | Explain the significance of the Ψ and Ψ^2 . | CO2 | L2 | 5M |

UNIT-II

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|---|---|--|-----|----|----|
| 4 | a | Compare the band diagrams of Insulators, Semi-conductors and Conductors. | CO2 | L3 | 5M |
| | b | Explain the basic principle and Classifications of Super Capacitors. | CO2 | L3 | 5M |

OR

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|---|---|--|-----|----|----|
| 5 | a | Draw the band diagrams for conductors, semi-conductors and Insulators. | CO2 | L2 | 6M |
| | b | Discuss the properties of Carbon nanotubes. | CO2 | L2 | 4M |

UNIT-III

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|---|--|---|-----|----|-----|
| 6 | | Derive the Nernst equation for a single electrode potential and explain the terms in equation and write its applications. | CO3 | L3 | 10M |
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OR

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|---|--|---|-----|----|-----|
| 7 | | Explain the construction, cell reactions and applications of Lithium-Ion rechargeable cell? | CO4 | L3 | 10M |
|---|--|---|-----|----|-----|

UNIT-IV

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|---|---|---|-----|----|----|
| 8 | a | Discuss the preparation, properties and application of Buna-S rubber and Buna-N rubber. | CO5 | L2 | 6M |
| | b | Explain about synthesis, properties and applications of Poly Lactic Acid. | CO5 | L2 | 4M |

OR

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|---|---|---|-----|----|----|
| 9 | a | Explain the preparation, properties and uses of Bakelite. | CO5 | L2 | 5M |
| | b | Explain Co-ordination or Ziegler-Natta polymerization. | CO5 | L2 | 5M |

UNIT-V

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|----|---|--|-----|----|----|
| 10 | a | What is meant by Chromatography? Write about principle and instrumentation of HPLC chromatography with neat diagram. | CO6 | L1 | 7M |
| | b | What is the use of detector in chromatographic technique and what are the different types of detectors used in HPLC technique. | CO6 | L2 | 3M |

OR

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|----|--|---|-----|----|-----|
| 11 | | Explain the various possible electronic transitions occurs in a molecule by absorbing the UV-Visible radiation. | CO6 | L2 | 10M |
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